



CAESAR: Modelling of the Sorption-Enhanced Water Gas Shift Process for Pre-combustion CO₂ Capture

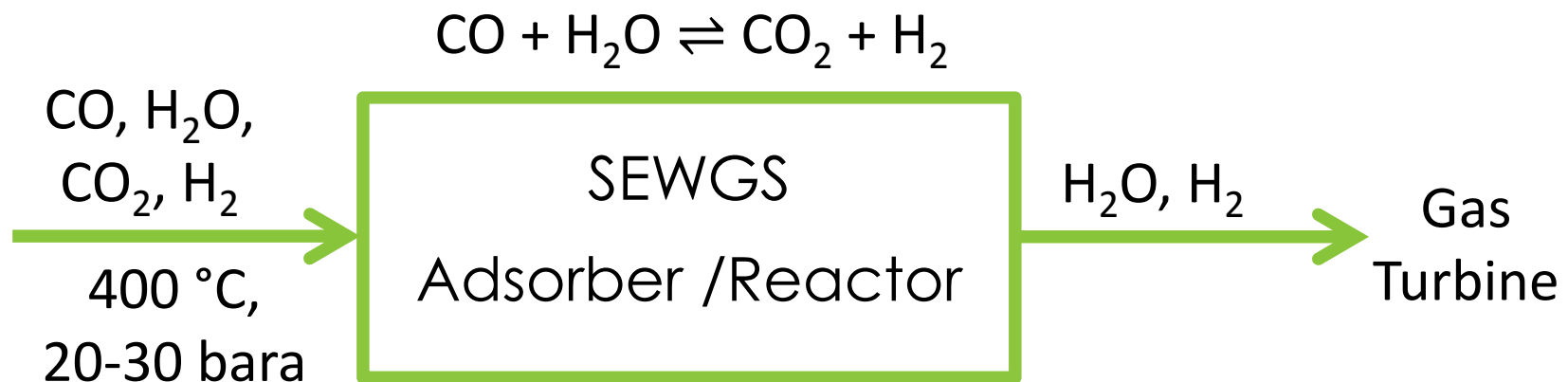
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What is SEWGS?

- Sorption-Enhanced Water-Gas-Shift (SEWGS)
- Combines the Water-Gas-Shift reaction with sorbent material to simultaneously produce H_2 at high temperature whilst also capturing CO_2

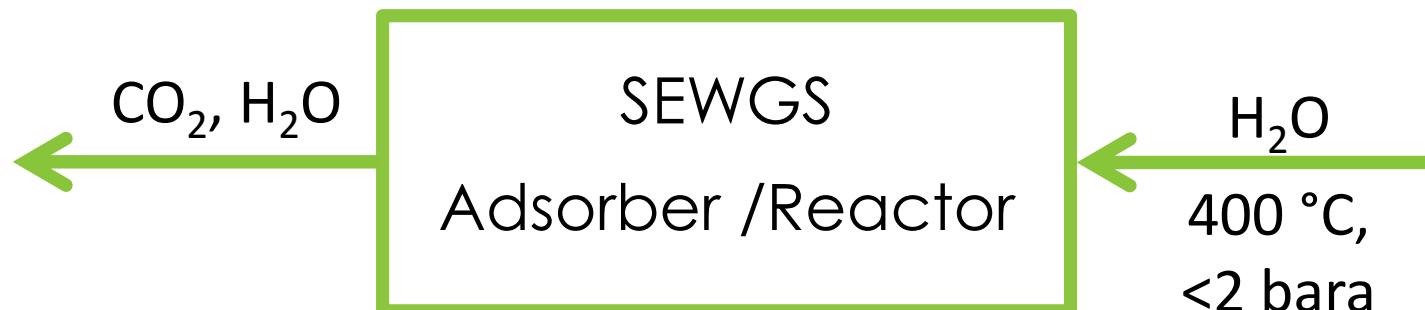


What is SEWGS?

- Before the CO₂ is regenerated from the sorbent, the vessel is **RINSED** using either steam or CO₂



- The CO₂ is regenerated from the sorbent by **PURGING** with low pressure steam



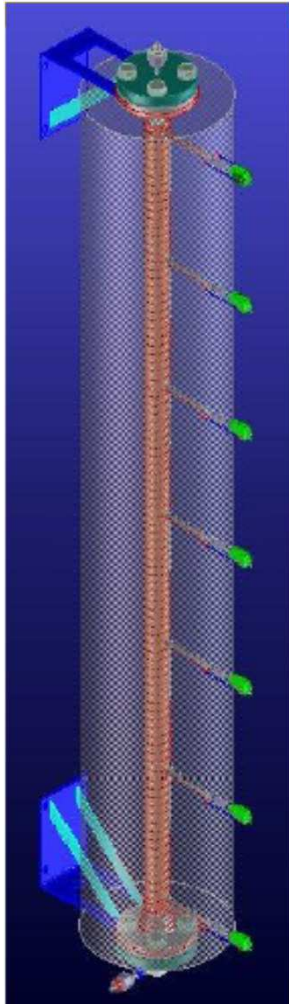
SEWGS and CAESAR

- Sorption-Enhanced Water Gas Shift (SEWGS)
 - Development started around 2000 by Air Products
 - CCP supported work
 - Material screening and technology feasibility study
 - CACHET
 - Construction and operation of test rig facilities
 - CAESAR
 - Material development, process optimisation, extension from natural gas to other feed-stocks
- CAESAR Consortium
 - Energy Research Centre of the Netherlands (ECN)
 - Air Products
 - SINTEF
 - Politecnico di Milano
 - BP

CAESAR Project

- Sorbent material used in previous studies was found not to be mechanically stable
- New material chosen for investigation in CAESAR
 - Physically more robust
 - Experimental data shows equivalent, if not better cyclic CO₂ capacity than original material
- Data collected on the sorbent material using various steady state and dynamic experiments
 - Gravimetric
 - Single column apparatus
 - Multi-column test rig

Test Rig Equipment



Single Column

2 metres tall
38 mm internal diameter

Multi-Column

6 vessels
6 metres tall
38 mm internal diameter



CAESAR: Scope of Work - Modelling

- Unfortunately, test rig data is not exactly representative of a full scale system
 - Heat management
 - Different numbers of vessels may be required
 - Different piping connections
 - Actual operating conditions may be different
- Therefore a full dynamic model of the SEWGS process is very useful for a number of reasons
 - Predict actual full-scale operation
 - Easier optimisation of the process
 - Evaluate benefit of further material development
- However model development is not a trivial task

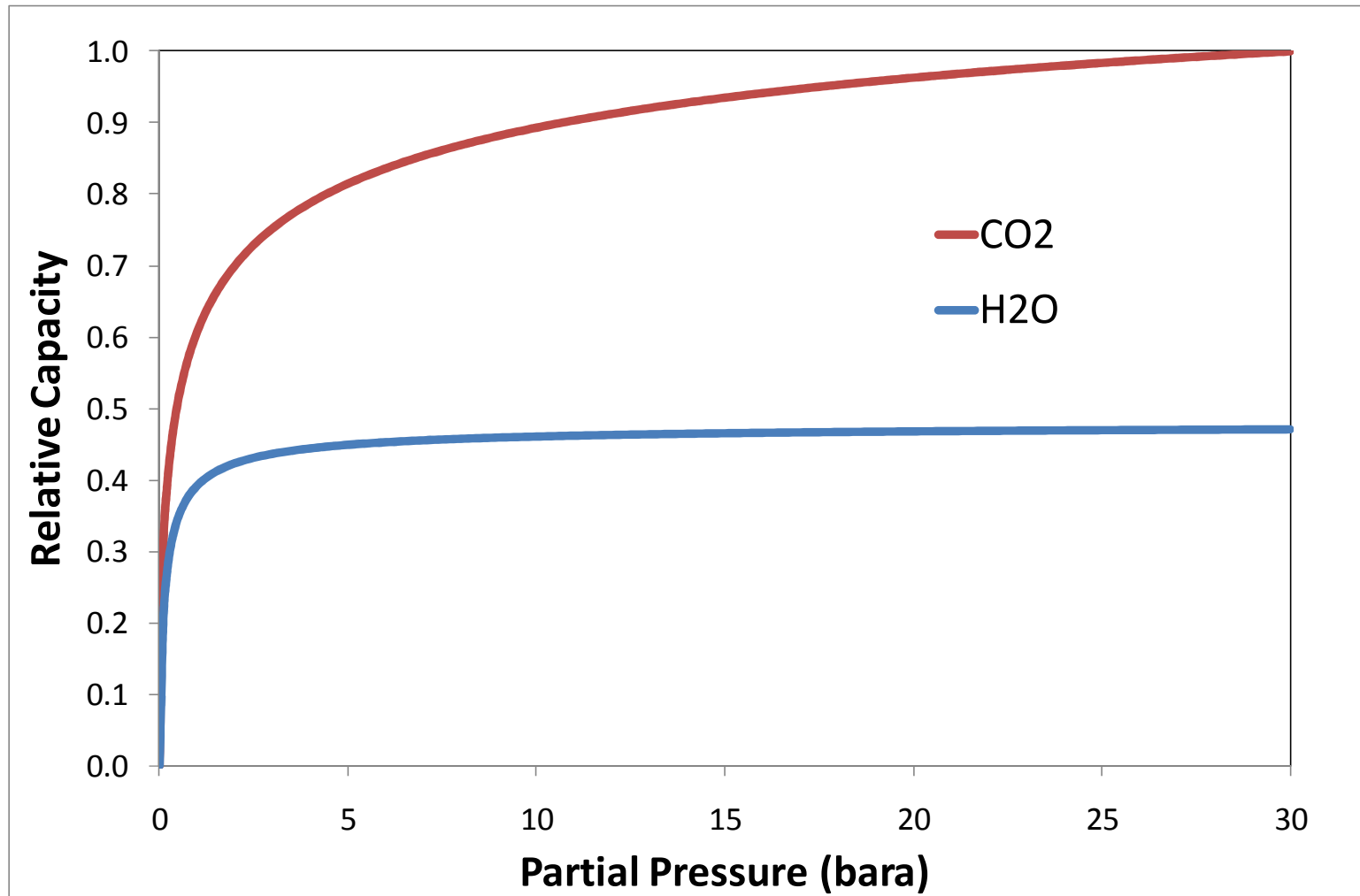
Adsorption Process Modelling

- Air Products has developed extensive modelling capabilities for adsorption processes over several decades
 - PSA – H₂ production
 - VSA – O₂ production
 - TSA – Air drying
- SEWGS has unique challenges
 - Much higher temperature than normal
 - 400 °C compared with typically 40 °C
 - Includes WGS reaction
 - Sorbent does not behave like traditional materials

Adsorption Process Modelling

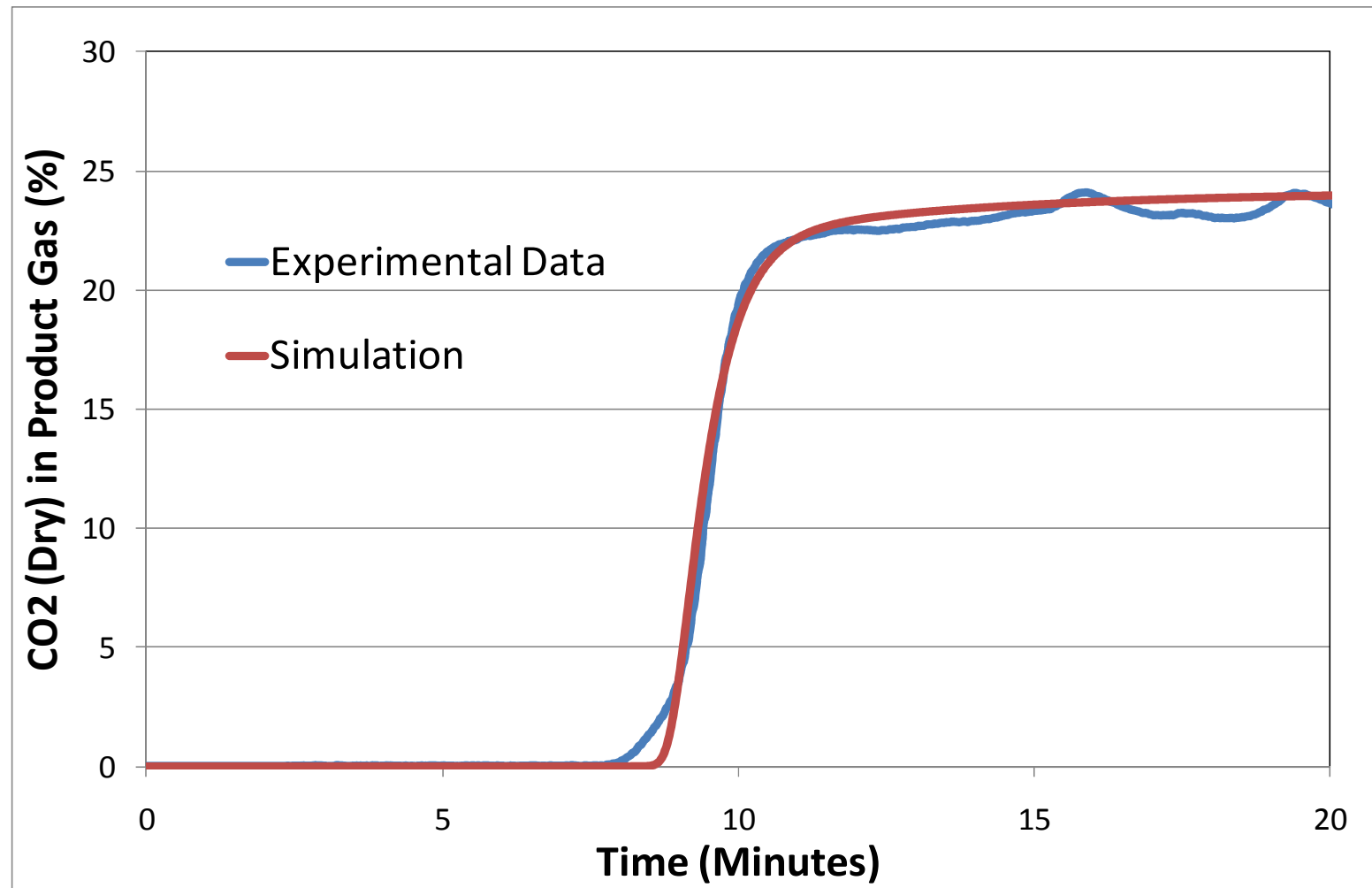
- Standard modelling building blocks still required
 - Overall heat and mass balance for the process
 - Material balances for all components
 - Momentum balance – i.e. Pressure drop
 - Equation of state (Ideal gas law)
- Additional challenges
 - Small test rig equipment has large pipe volumes
 - Inclusion of reaction terms
 - Assumed always close to equilibrium
 - Accurate model of the sorbent required

Pure Component Isotherms (400 °C)



- Most of the capacity change is at low partial pressures

Breakthrough Experiment (400 °C)



- Adsorption mass transfer resistance of <1 second

Single Column Cyclic Experiments

Experimental Test Case	Cycle-Average Mass Transfer Resistance (s)
Steam-Rinse Test 1	95.2
Steam-Rinse Test 2	94.3
Steam-Rinse Test 3	120.5
Steam-rinse Test 3 (Repeat)	126.6
CO ₂ -Rinse	109.9

- Rate of desorption is much slower than adsorption

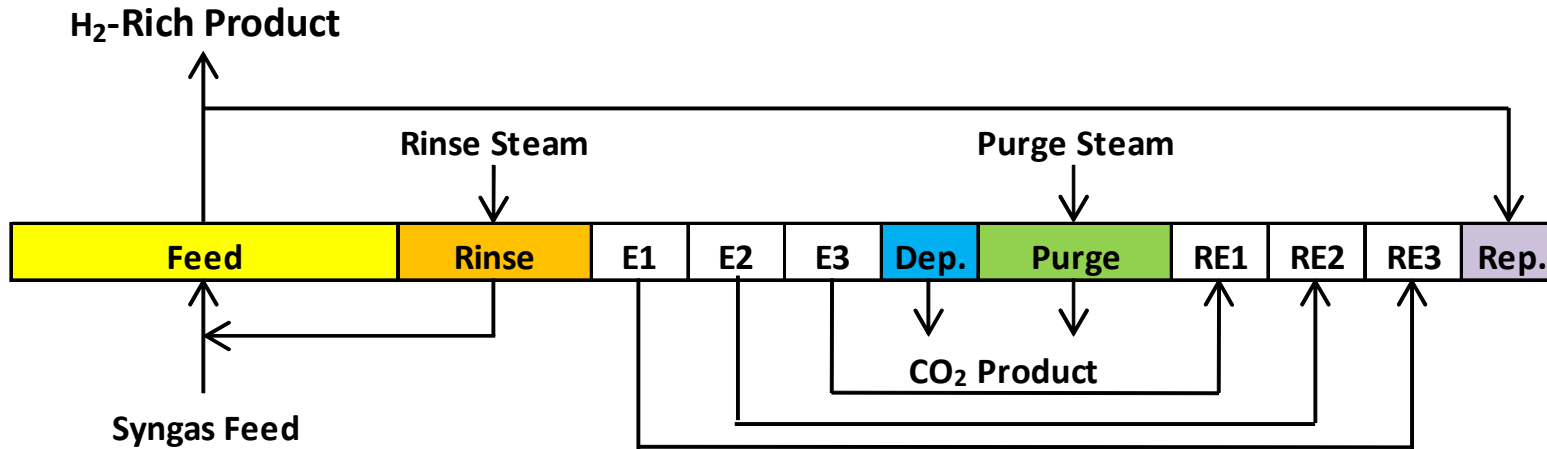
Full-Scale Design

- Example natural gas and coal based feed-stocks

	Natural Gas	IGCC
CO ₂ (%)	12	24
CO (%)	5	6
H ₂ (%)	42	35
H ₂ O (%)	8	31
N ₂ (%)	32	3
Others (%)	1	1

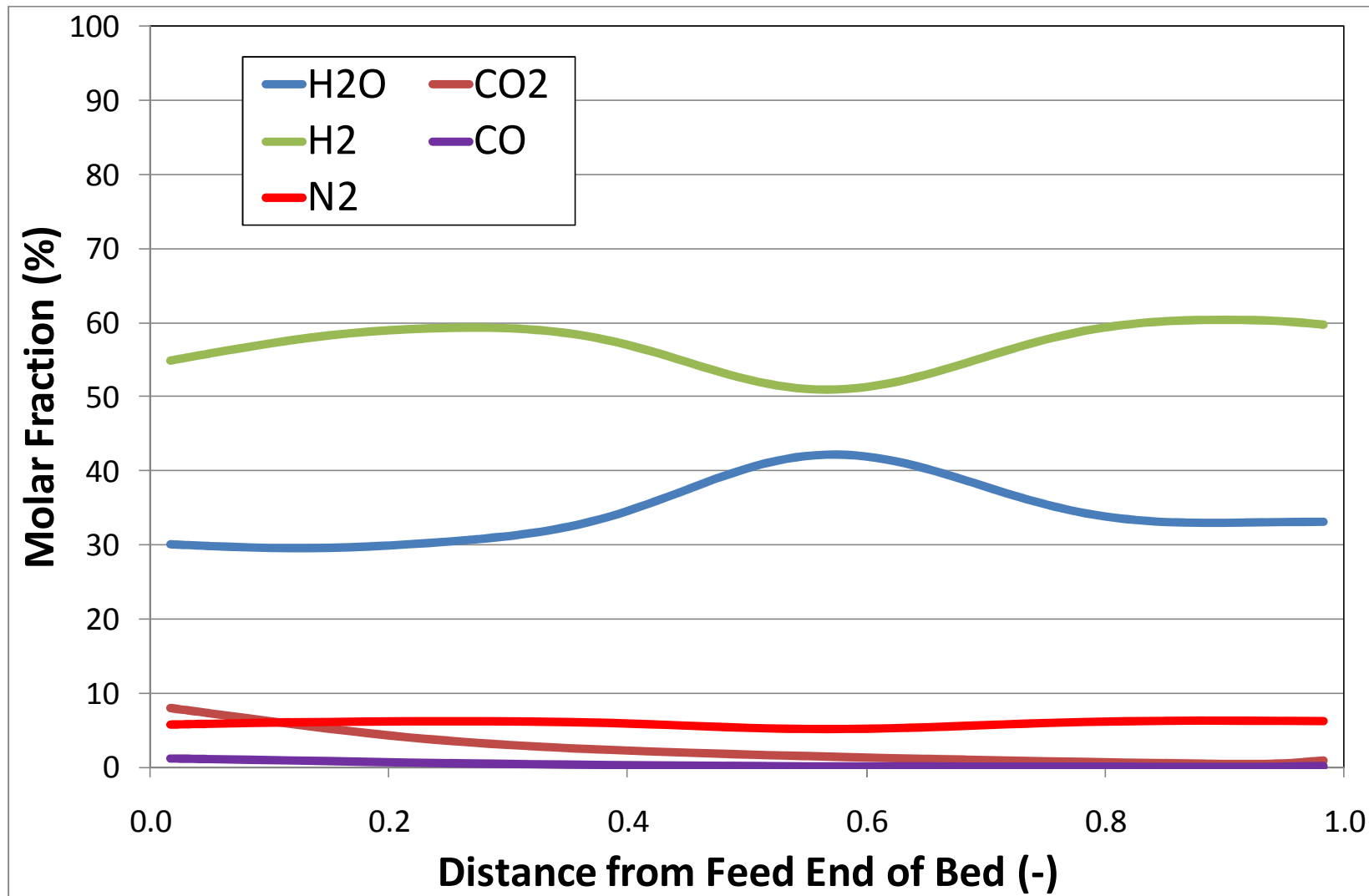
- Optimisation parameters available for investigation
 - Feed conditions, Bed length, Number of trains, Cycle time, CO₂ capture rate, CO₂ purity, Purge pressure, Cycle sequence, Sorbent physical properties, etc...

Full-Scale Design – Example Cycle

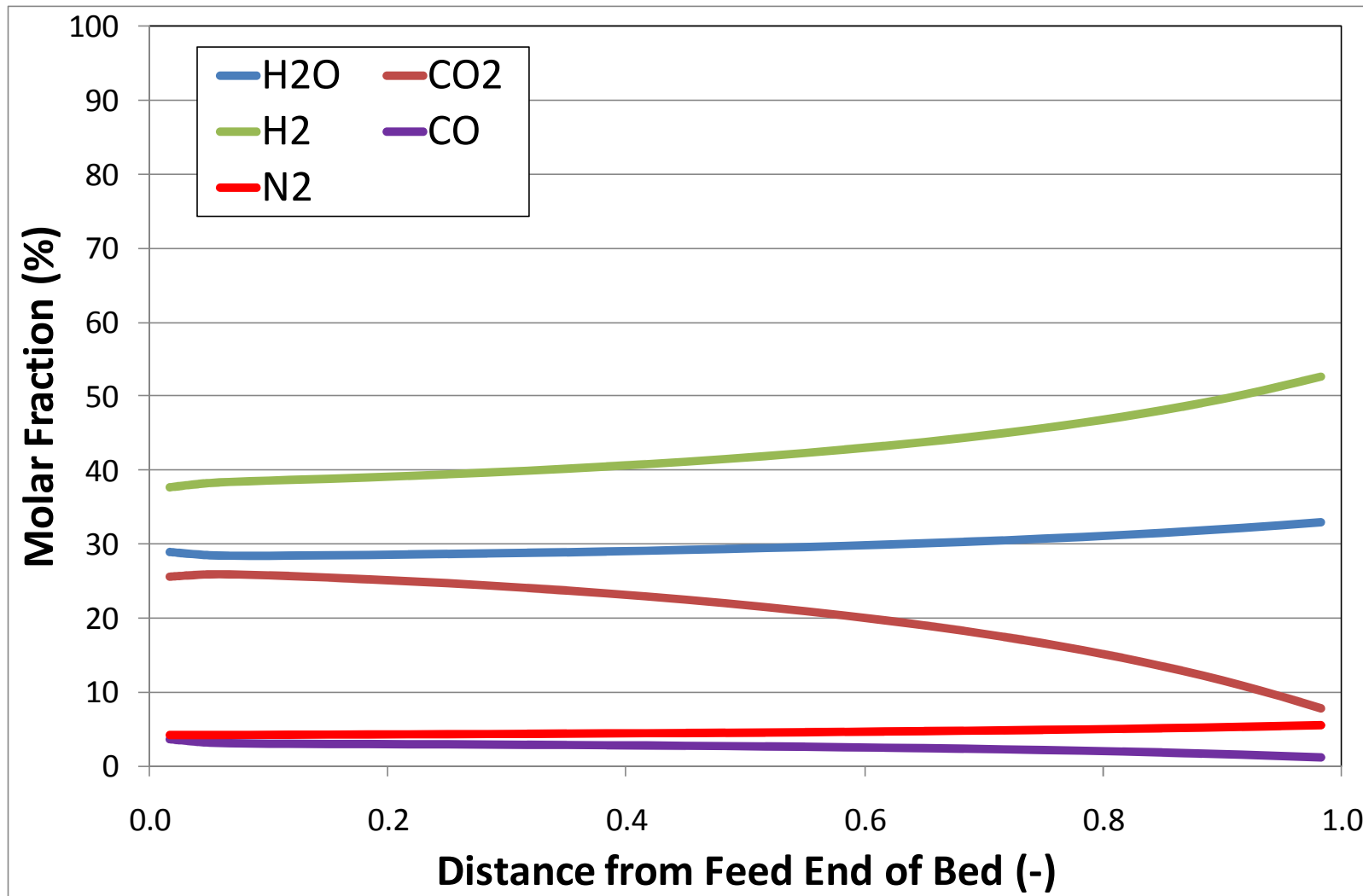


1	Feed		Rinse	E1	E2	E3	Dep.	Purge	RE1	RE2	RE3	Rep.
2	RE3	Rep.	Feed		Rinse	E1	E2	E3	Dep.	Purge	RE1	RE2
3	RE1	RE2	RE3	Rep.	Feed		Rinse	E1	E2	E3	Dep.	Purge
4	Purge	RE1	RE2	RE3	Rep.	Feed		Rinse	E1	E2	E3	Dep.
5	E3	Dep.	Purge	RE1	RE2	RE3	Rep.	Feed		Rinse	E1	E2
6	E1	E2	E3	Dep.	Purge	RE1	RE2	RE3	Rep.	Feed		Rinse
7	Rinse	E1	E2	E3	Dep.	Purge	RE1	RE2	RE3	Rep.	Feed	
8	Feed	Rinse	E1	E2	E3	Dep.	Purge	RE1	RE2	RE3	Rep.	Feed

Example Cycle Modelling – Start of Feed

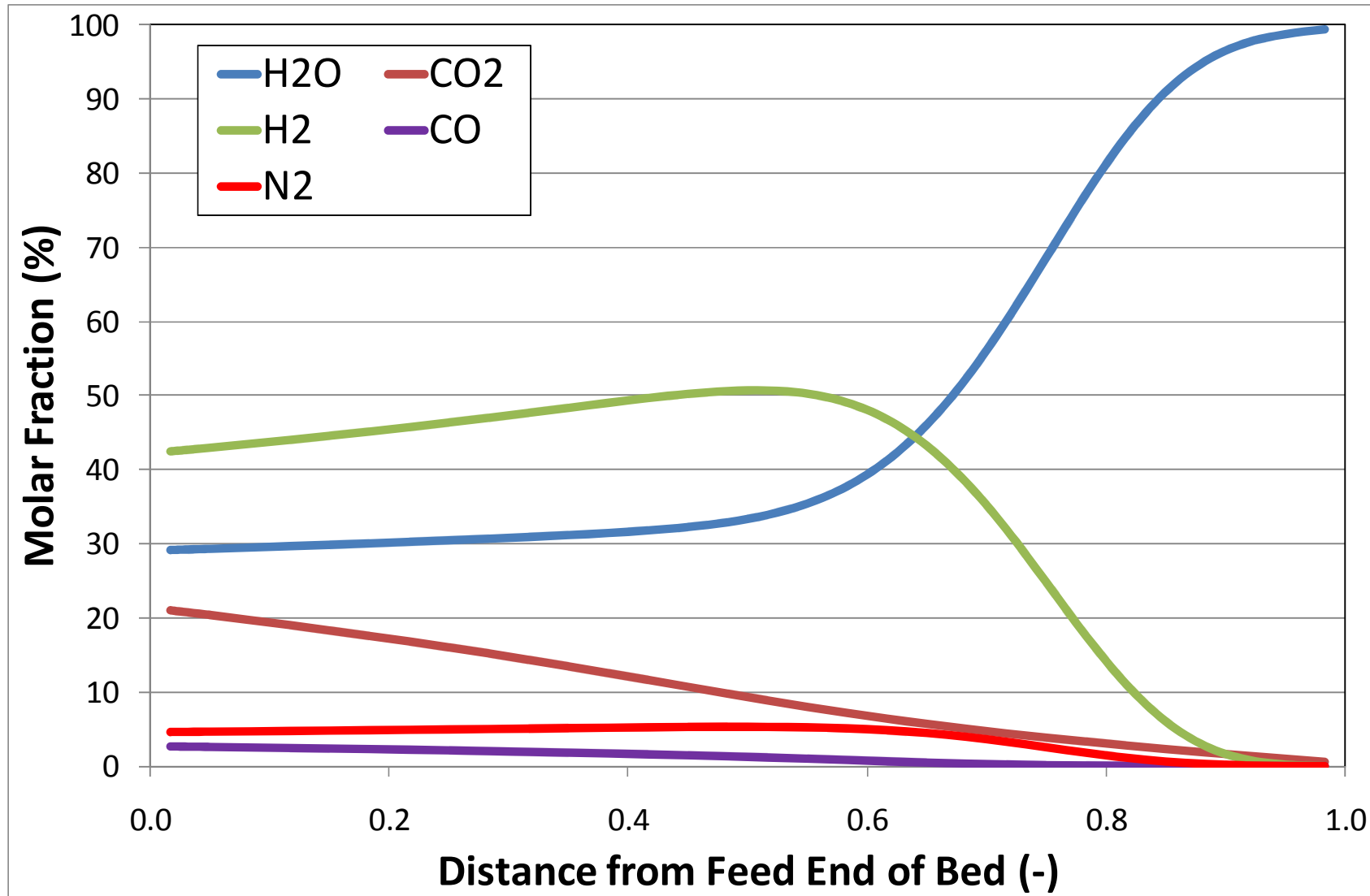


Example Cycle Modelling – End of Feed

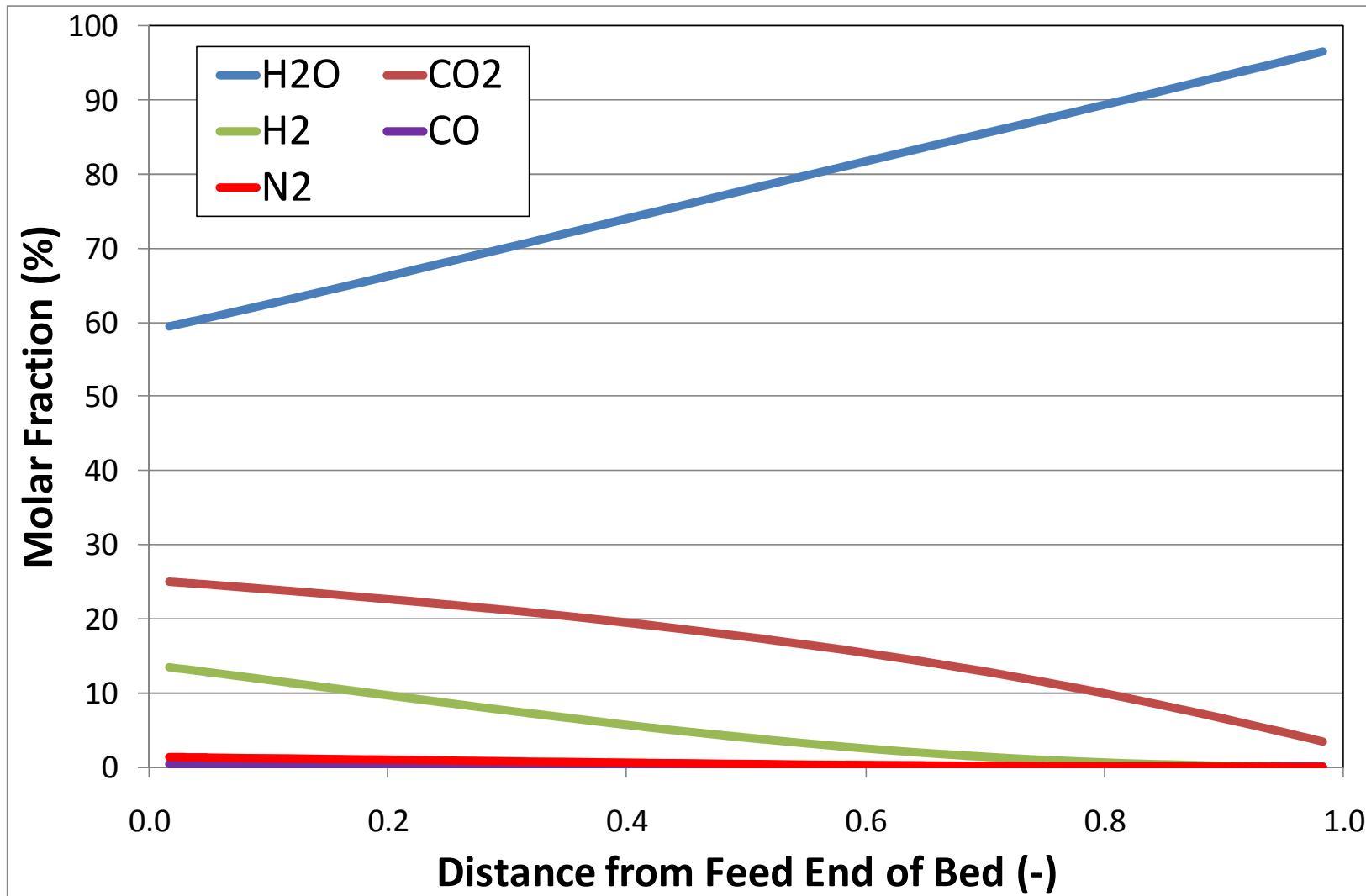




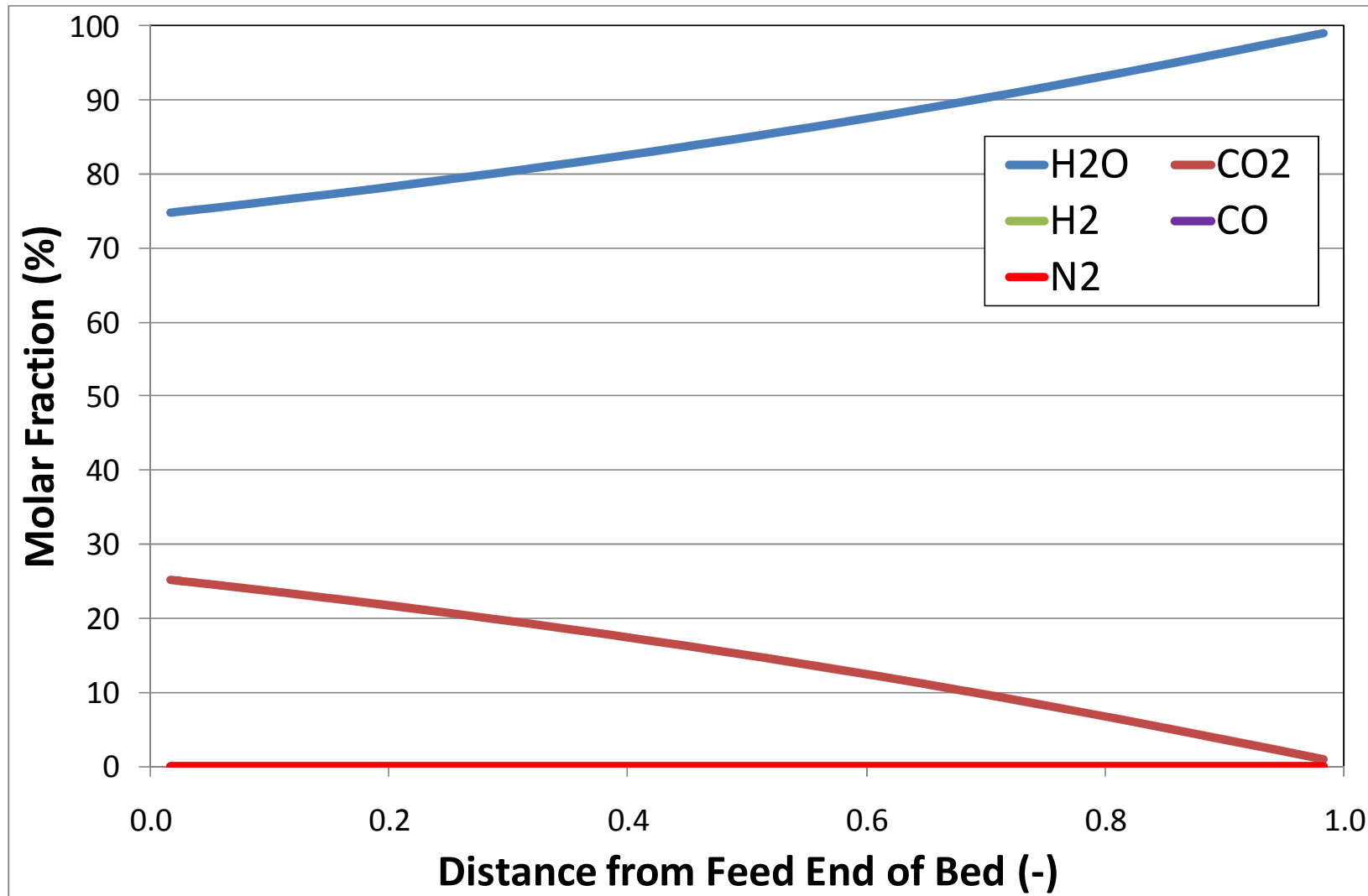
Example Cycle Modelling – Start of Rinse



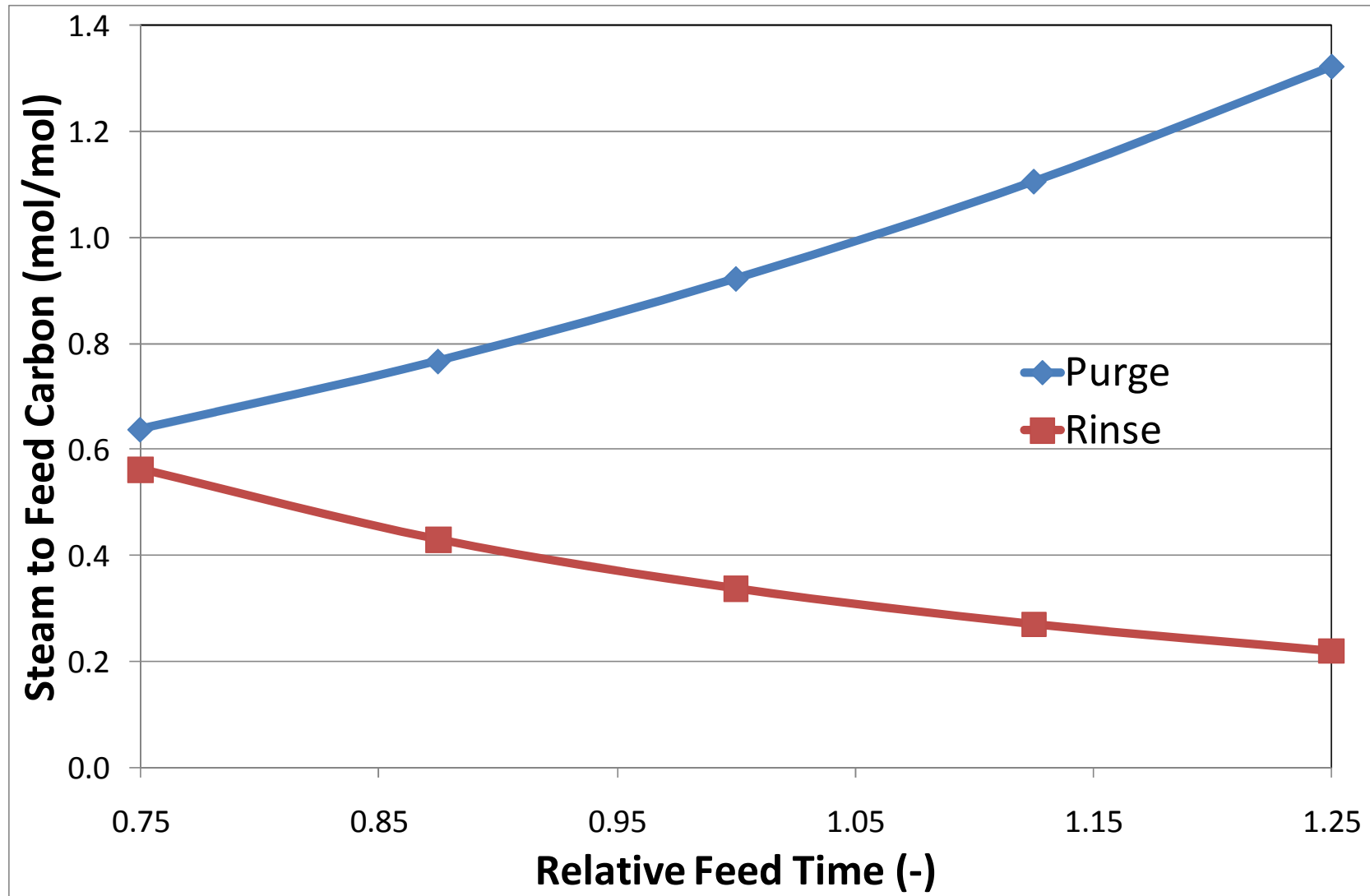
Example Cycle Modelling – End of Rinse



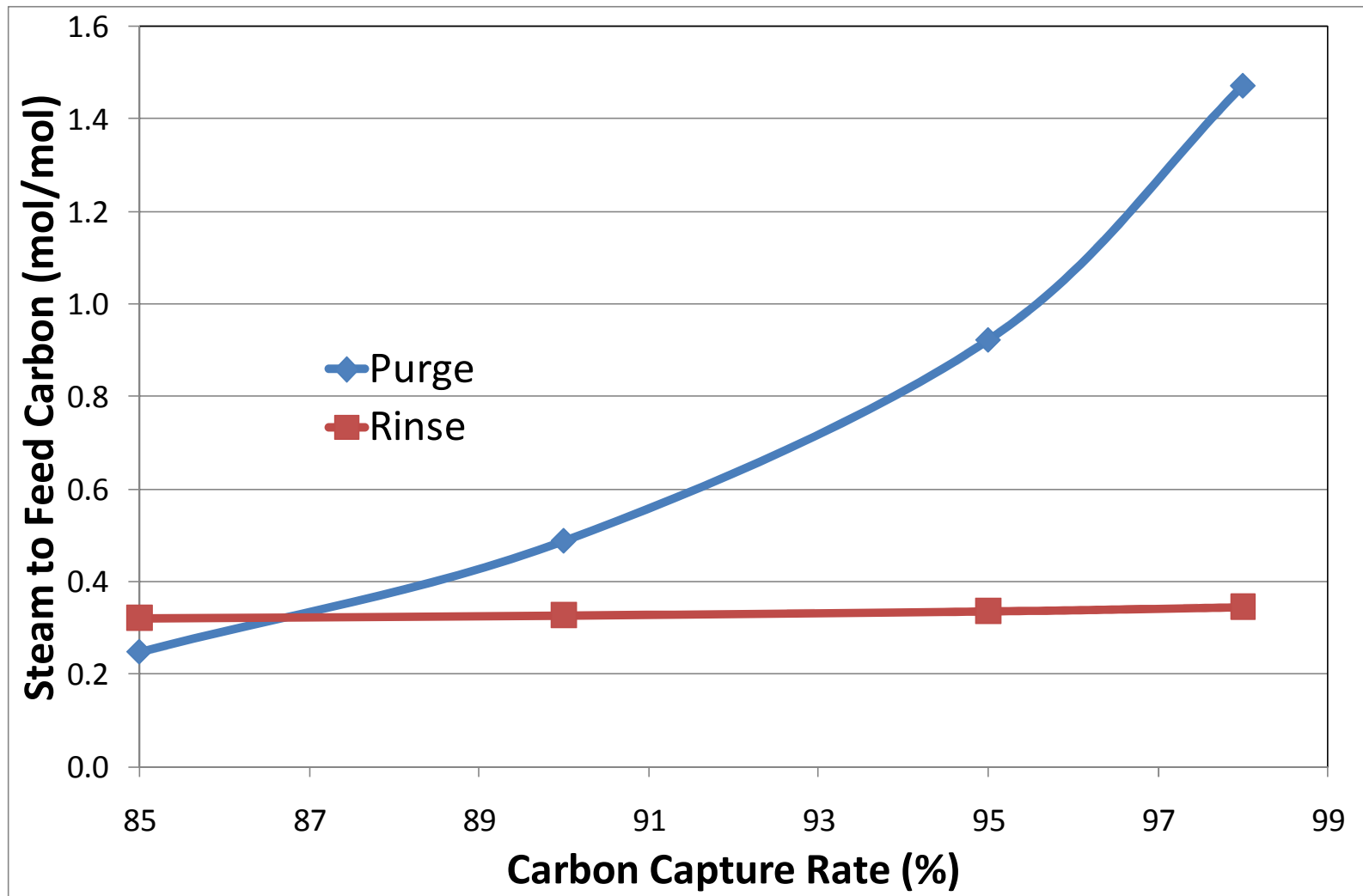
Example Cycle Modelling – End of Purge



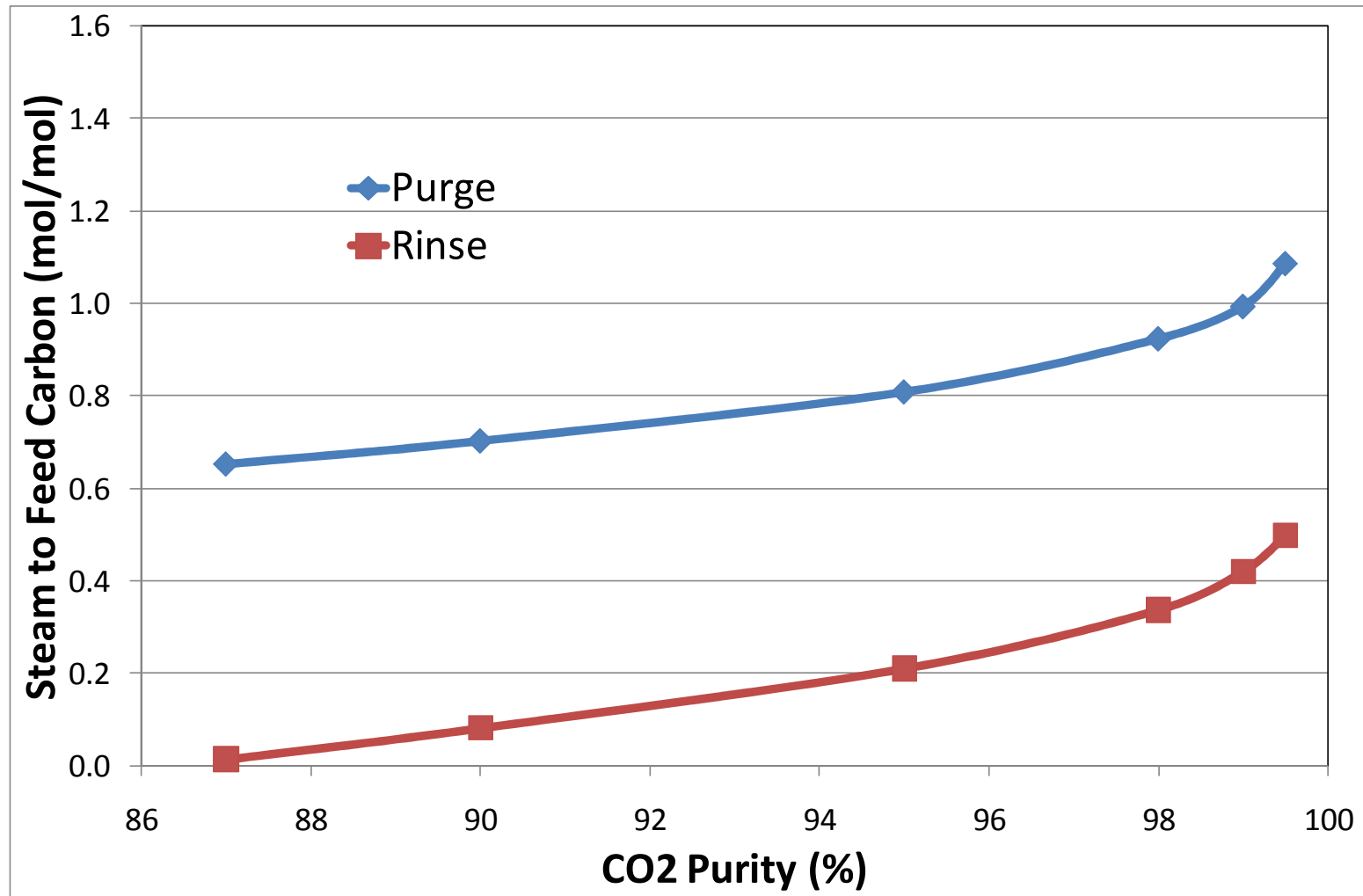
Cycle Time – IGCC Case



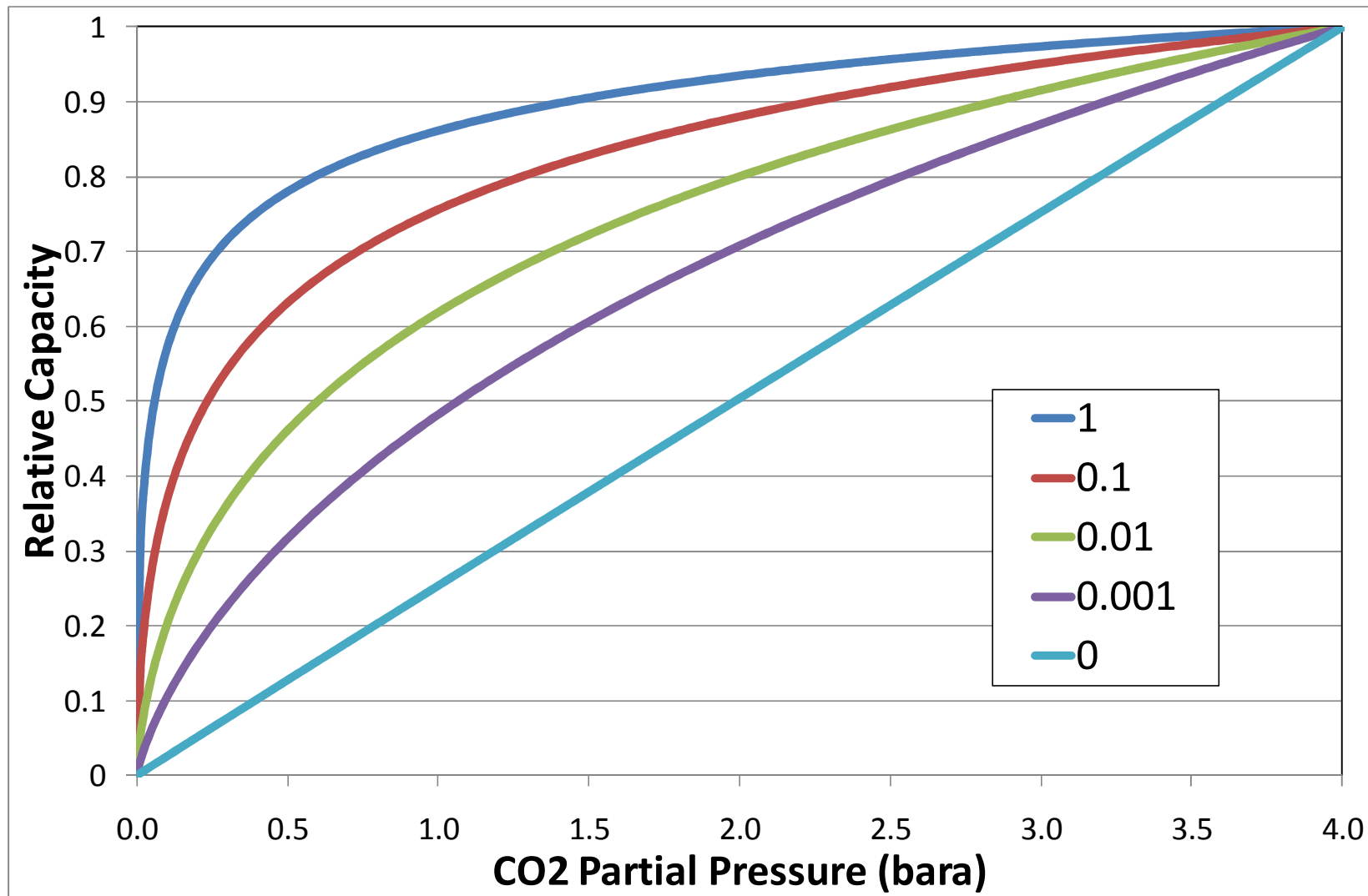
Carbon Capture Rate – IGCC Case



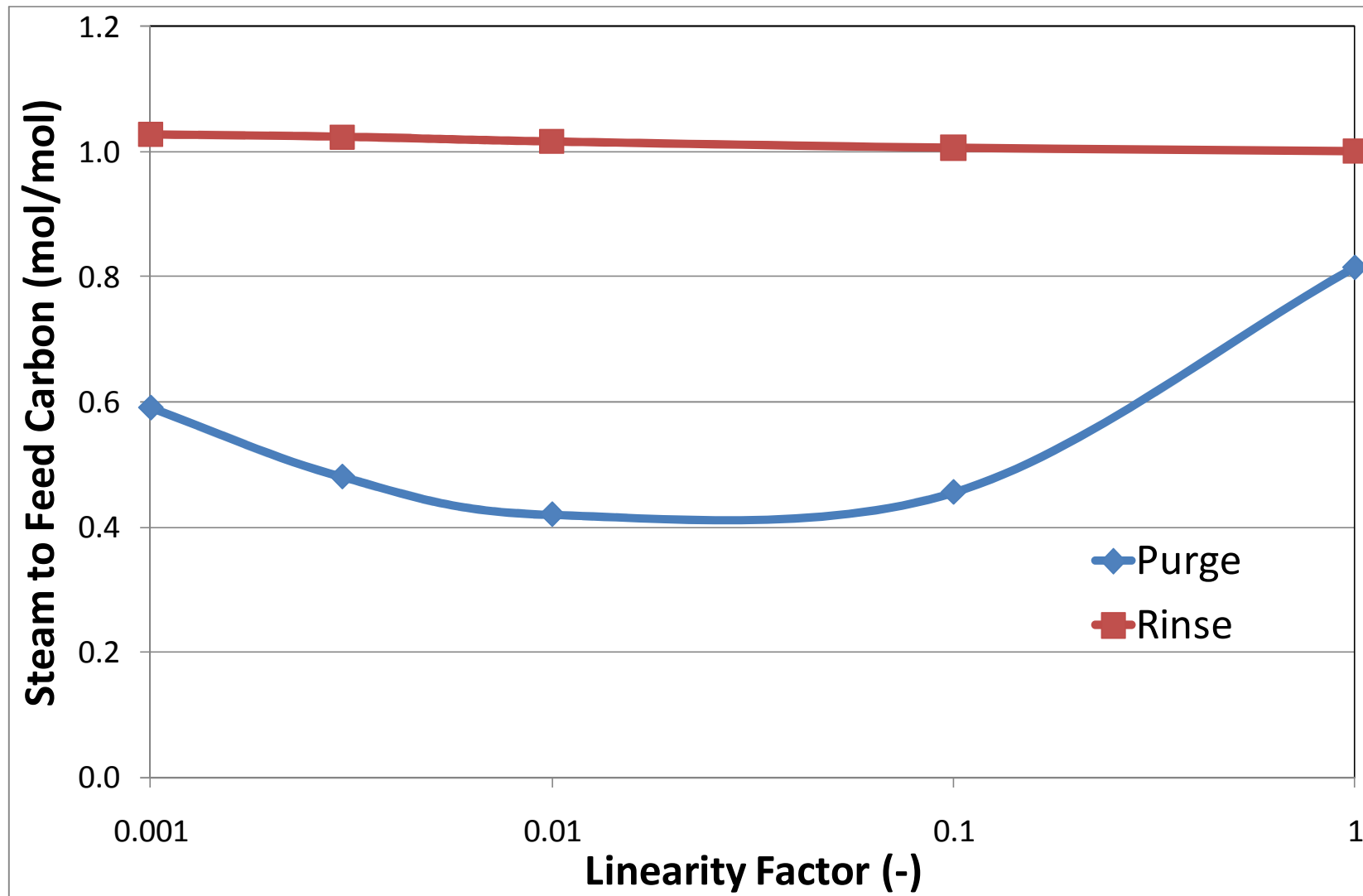
CO₂ Purity – IGCC Case



Material Development – Isotherm Shape



Linearity Factor – Natural Gas Case



Summary

- Within CAESAR, a model has been created for the SEWGS process based on a new sorbent material
- Predicted performance versus breakthrough and cyclic experiments in a single column unit show reasonable agreement over a range of conditions
- This model has been used to estimate the performance of a full-scale unit to evaluate changing different operating parameters
- Interesting results in regard to improved material formulation have arisen
- Next step is to optimise the SEWGS process as part of the entire power plant flow sheet
 - Preliminary results are encouraging

Summary

- The model needs further validation and to be improved based on cyclic data obtained from a larger multi-column test rig at ECN
- Key areas for improved modelling are:
 - A better understanding on kinetic versus equilibrium constraints affecting the rate of desorption of CO₂
 - Improved modelling of connecting pipes (particularly if high purity CO₂ is targeted)



Thank you

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